## ANNEXURE-IV (Bridge Course)

Programme: M.Sc. Chemistry

## Course Code: CHCB-401 (for Part-I students)

Title of the course: Bridge Course in mathematical concepts for chemistry

Number of Credits: 01 Total Hours: 15 Effective from AY: 2022-23

| Prerequisites<br>for the course.   | Should have studied B. Sc. (Chemistry)                                  |      |  |
|--|---|------|--|
| Course   | To introduce mathematical concepts to the students of MSc Part-I        |      |  |
| Objective:   | (Chemistry).  |      |  |
| Course   | Students will be able to solve problems based on matrices, determina    | ints |  |
| Outcome:   | and, differential and integral calculus in MSc Chemistry.               |      |  |
|  | Content   | Hrs  |  |
| 1. Calculus fo   | r thermodynamics and kinetics   | 08   |  |
| i. Introduction  | to Differentiation: Notation, Differentiating various functions,        |      |  |
| Differentiating  | Differentiating a Sum, Product Rule, Quotient Rule, Chain Rule, Partial |      |  |
| Differentiation: exact and inexact differentials.  |   |      |  |
| ii. Introduction to Integration: Notation, Rules for Integrals, Integrating various      |   |      |  |
| functions, Def   | inite and indefinite Integrals.   |      |  |
| 2. Matrices, Determinants and vector algebra:  |   | 07   |  |
| i. Types of Matrices: Identity, reflection, rotation, inversion, distance matrix, Matrix |   |      |  |
| Algebra, Matrix similarity transformation.   |   |      |  |
| ii. The Determinant, Minors and Cofactors, Inverse of a Matrix, Character of a           |   |      |  |
| matrix, Linear algebra.  |   |      |  |
| iii. Vectors and molecular structure.  |   |      |  |
| Pedagogy   | Mainly lectures and tutorials. Seminars/assignments/presentations/self  | -    |  |
|  | study or a combination of some of these can be used. ICT mode should    | d be |  |
|  | preferred. Sessions should be interactive to enable peer group learning |      |  |
| Text Books/  | 1. Robert G. Mortimer, Mathematics for Physical Chemistry, Elsevier,    | ,    |  |
| References /   | 2013, 4 <sup>th</sup> Ed.   |      |  |
| Readings   | 2. James R. Barrante, Applied Mathematics for Physical Chemistry,       |      |  |
|  | Prentice-Hall, 1998, 3 <sup>rd</sup> Ed.                                |      |  |